

Cube Root

Ex. 244,140,625 mark off by 3 digits at one time.

244	140	625		625	(a)	(b)
216					182	10800
28140						364
22328						11164
5812625					1865	4
5812625						1153200
-----						9325
						1162525

Temperature of water = 4°C (39.2°F)

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Steam at 100°C is 1600 times the bulk of water at 100°C or 1650 times the bulk of water at 4°C

All gases expand $\frac{1}{273}$ for every 1°C rise in temperature

(5370)
540° units of heat (each sufficient to raise 1 lb. of 1°C) are required to convert a pound of water into steam

Heat necessary to raise one gram of water from 0° to 1°C is called a unit of heat or 1 calorie

-273°C or -460°F are supposed to be devoid of heat

Absolute temperature of boiling water equals $212^\circ\text{F} + 460^\circ = 672$ (at atmospheric pressure)

772 ft. lbs. = 1 Unit of heat lbs. F°
1390 ft. lbs. = do lbs. C°

$$\frac{33000}{772} = 42.75 = 1 \text{ HP}$$

Absolute is pressure as ordinarily indicated by the pressure gauge + atmospheric pressure

966 units of heat are required to evaporate 1 lbs. of water at 212°F to 1 lbs. of steam at 212°F

Latent heat of steam at a given pressure, $= 1114 - .7t$ when $t =$ temperature at that pressure

Total heat of evaporation from steam at any particular temperature = $1082 + .3t$ approx

$P \times 144 \times V =$ work done per lb. of steam
when $P =$ absolute pressure (lb. sq. in.)
 $V =$ volume of 1 lb. in cu. ft.

1 lb. of good coal give 14,500 units
of heat on combustion $778 \times 14,500$
= 10,808,000 units of work

1 cubic foot of water weighs $62 \frac{3}{4}$ lbs

1 gallon weighs 10 lbs

1 cubic foot = $6 \frac{1}{4}$ gallons

1 ton = 224 gallons

1 cu ft = 1000 ozs.

1 gallon = 277.274 cu ins

To find the weight of steam
required per horsepower per hour,
- divide work done per horse
power per hour by work done
per lb. of steam

Ex. Work per HP/hr = $33,000 \times 33$
= 1,980,000 ft. lbs. Work done per
lb. of steam at 100 lb p.s.a.

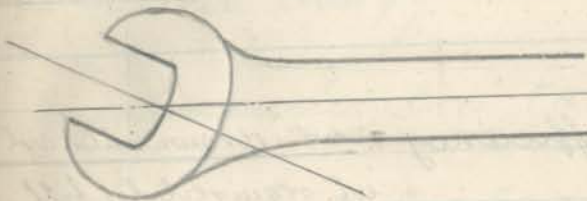
= $100 \times 144 \times 4.33 = 62,352$ ft. lbs.

$\therefore \frac{1,980,000}{62,352} = 31.7$ lbs. = The weight

of steam required per horsepower
per hour under the above
conditions

Algebraic Fractions:-

$$\frac{1}{2-x} = \frac{-1}{x-2} = -\frac{1}{x-2}$$



Total efficiency = Air standard efficiency
 x relative efficiency x mechanical eff.

The A.S.E. & R.E. are usually taken together and called the Indicated Efficiency

Volumetric efficiency = $\frac{\text{wt. drawn into cyl}}{\text{wt. required to fill at atmospheric pressure}}$

3 causes of mechanical loss are:-

- 1) Piston friction
- 2) Bearing friction
- 3) Fluid friction

Torque, etc

$$HP = \frac{2\pi N \times M}{33,000}$$

$$M = \frac{33,000 \times HP}{2\pi N}$$

where $N = \text{R.P.M}$ & $M = \text{torque in ft. lbs.}$

In twisting an elastic rod, work done
 = $\frac{1}{2} M \theta$ where $\theta = \text{angle of twist}$
 in radians & $M = \text{maximum turning}$
 moment also in compressing a spring
 etc Work = $\frac{1}{2} ML$

$M = \text{max. force applied in lbs.}$
 $L = \text{distance in ft.}$

$$K.E. = \frac{W V^2}{2g}$$

work done = $\frac{W}{2g} (v^2 - u^2) = \text{change}$
 of kinetic energy

Angular acceleration $FR = I \frac{\omega}{t}$

turning moment = $FR = \frac{\omega R^2 M}{t \cdot 32.2}$

First moment = MR

Second moment = $\frac{MR^2 \omega}{t \cdot 32.2}$

Centrifugal Force = $\frac{WN^2 r}{2935}$

$W = \text{weight in lbs.}$

$N = \text{revs. per min}$

$r = \text{radius of circle in feet}$

Road Springs

The length and breadth of springs must be decided and the load to carry assumed

Then
$$N = \frac{1.5WL^2}{fbt^2}$$

where N = no. of plates
 f = stress
 b = breadth in inches
 t = thickness in deg
 L = span in deg
 W = load in lbs.

For this calculation the thickness also can be assumed and adjusted afterwards to give correct period of oscillation

Deflection

$$\Delta = \frac{W(L-l)^3}{42000 n t^3 b}$$

Δ = deflection
 W = load in tons
 l = distance over clips

Total eff
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 together
 Efficien

Volumetr

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 1)
 2)
 3)

Torque. etc

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In twisting
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 in radial
 moment
 etc Wa